

# High Performance Flexible Reversible Solid Oxide Fuel Cell

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imagination at work

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# Overview

## Timeline

- Project start date: October 2004
- Project end date: September 2006
- Percent complete: 75%

## Budget

- Total project funding
  - DOE share: \$1,252,683
  - Contractor share: \$616,993
- Funding received in FY05: \$575,198
- Funding for FY06: \$677,485

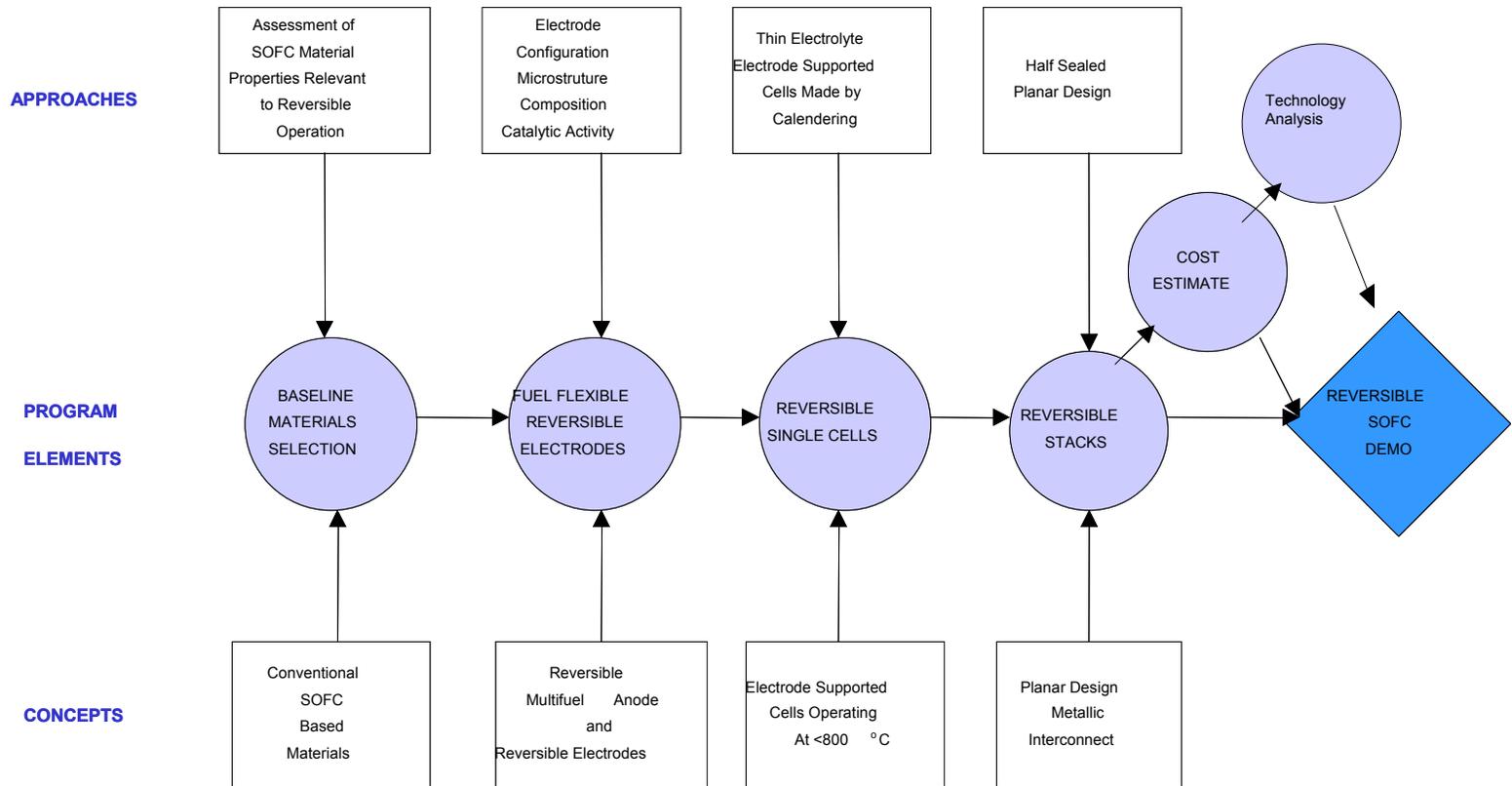
## Barriers

- Barriers addressed
  - K. Electricity Costs
  - G. Capital Costs
  - H. System Efficiency

# Objectives

- Demonstrate a single modular stack that can be operated under dual modes
  - Fuel cell mode to generate electricity from a variety of fuels
  - Electrolysis mode to produce hydrogen from steam
- Provide materials set, electrode microstructure, and technology gap assessment for future work

# Approaches



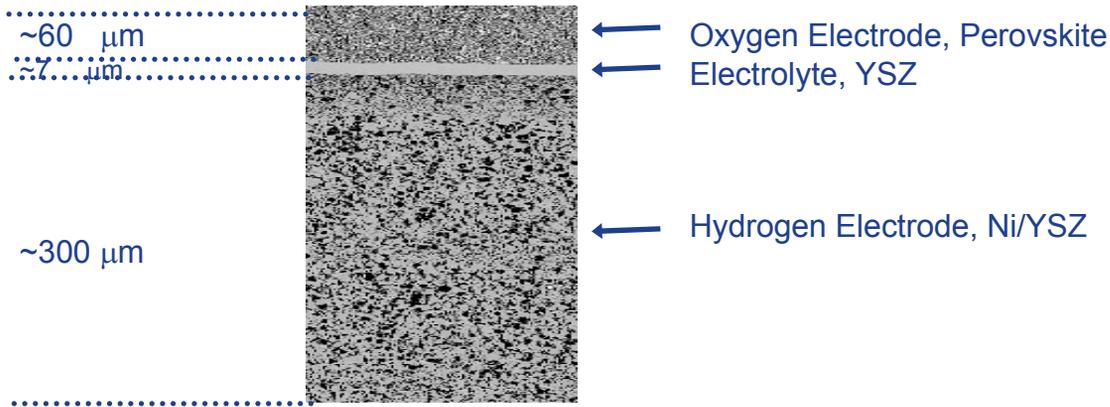
## Technical focuses:

- Reversible electrode modeling
- Electrode compositions and microstructure engineering

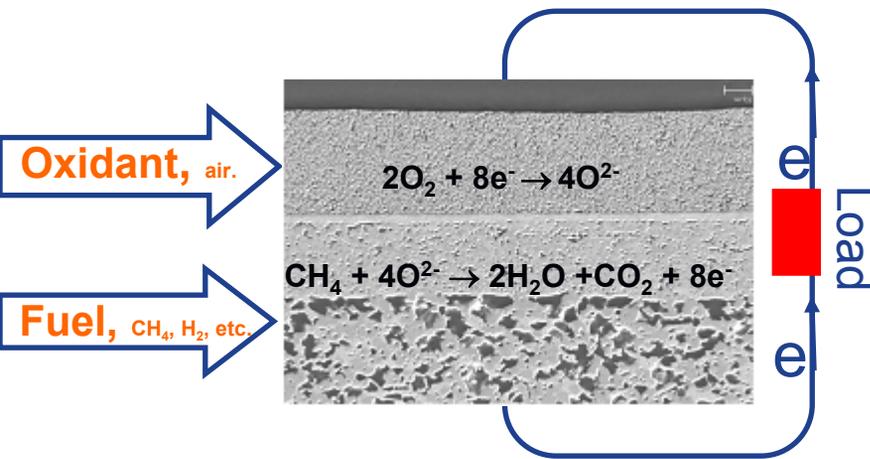
## Key challenges:

- Performance for cost and efficiency
- Low degradation for reliability

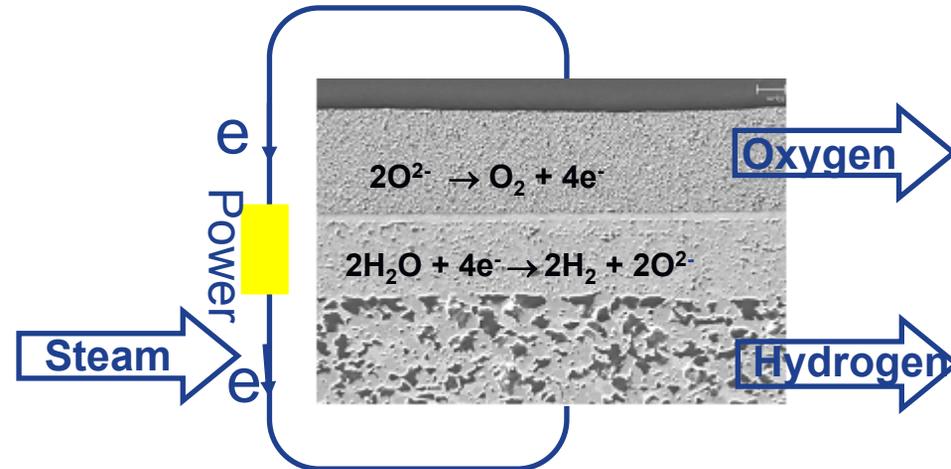
# Cell Configuration



## Power Generation Mode



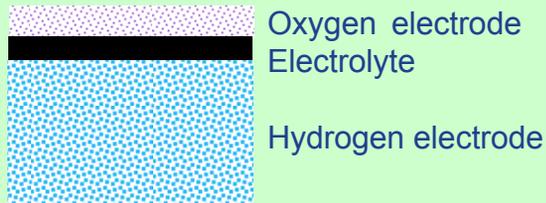
## Hydrogen Production Mode



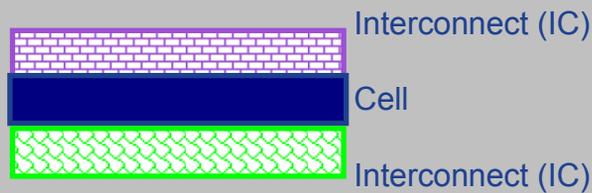
- SOFCs have the flexibility, running under power generation mode and hydrogen production mode
- High temperature solid oxide steam electrolysis can lower the electricity consumption

# Stack Configuration

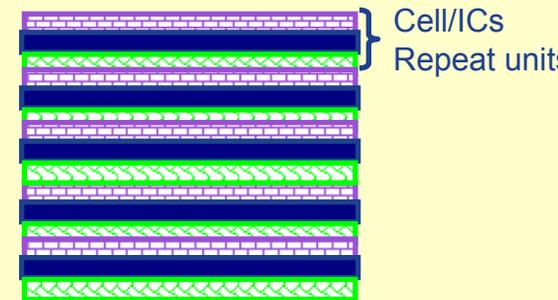
## Cell



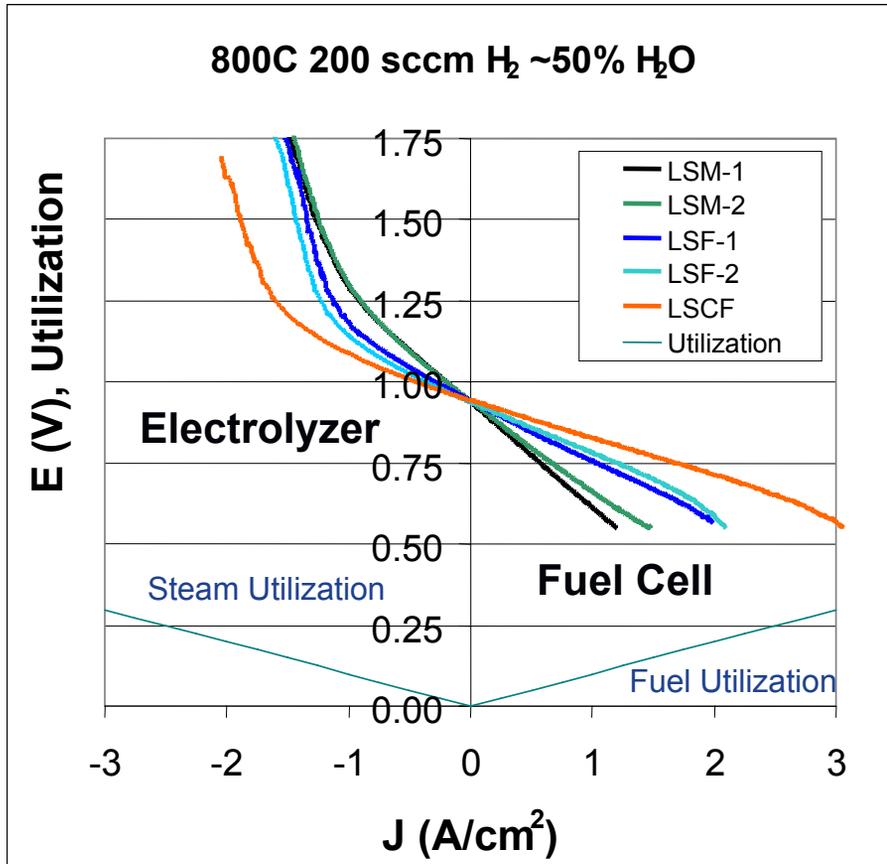
## Module



## Multi-cell Stack

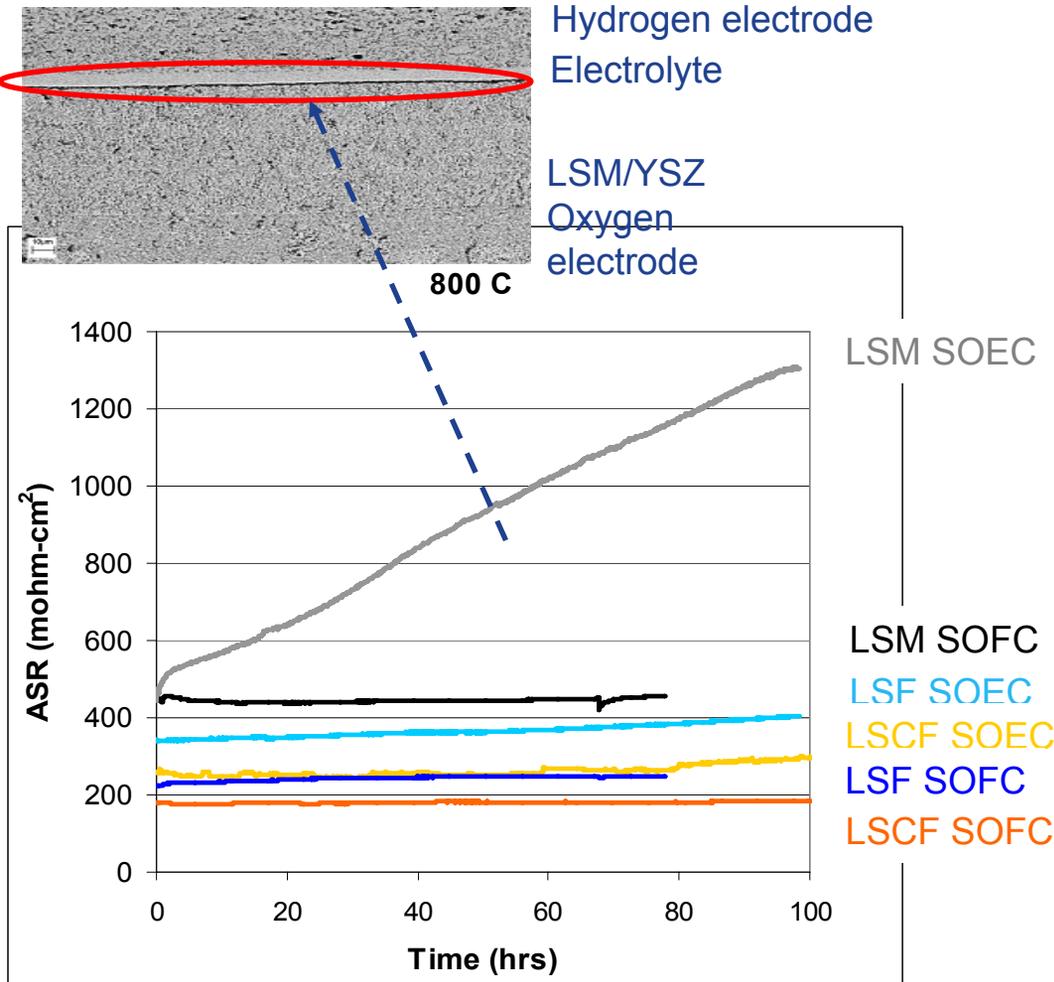


# Oxygen Electrode Performance



- Screened several lanthanum strontium manganites (LSM), lanthanum strontium ferrites (LSF), and lanthanum strontium cobalt iron oxides (LSCF) as oxygen electrodes
- Under both modes, electrode performance increases in the order of LSCF>LSF>LSM/YSZ

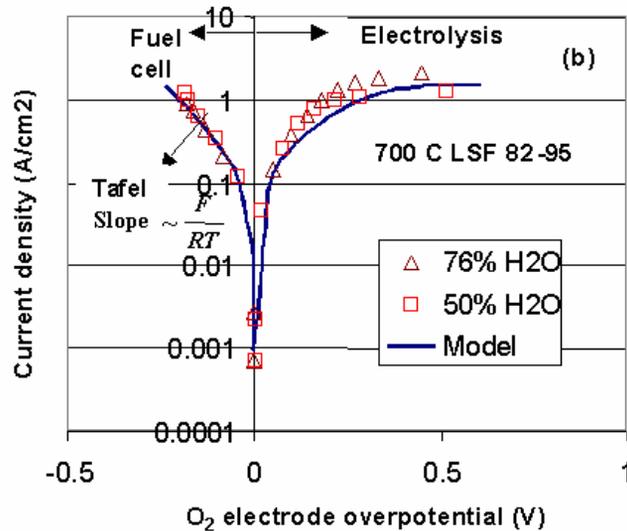
# Oxygen Electrode Performance Stability



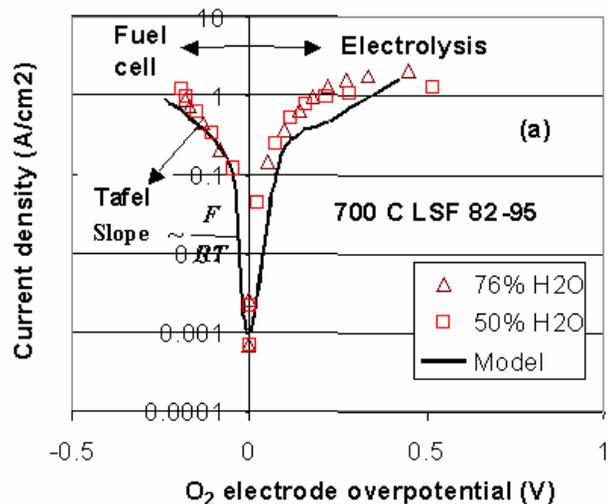
- **Excess performance degradation was observed with LSM/YSZ as the oxygen electrode in electrolysis mode (SOEC) mainly due to electrode delamination**
- **LSCF and LSF showed better performance stability in electrolysis mode than LSM/YSZ electrode**

# Oxygen Electrode Reversibility

Non-symmetrical vacancy model



Non-symmetrical vacancy model

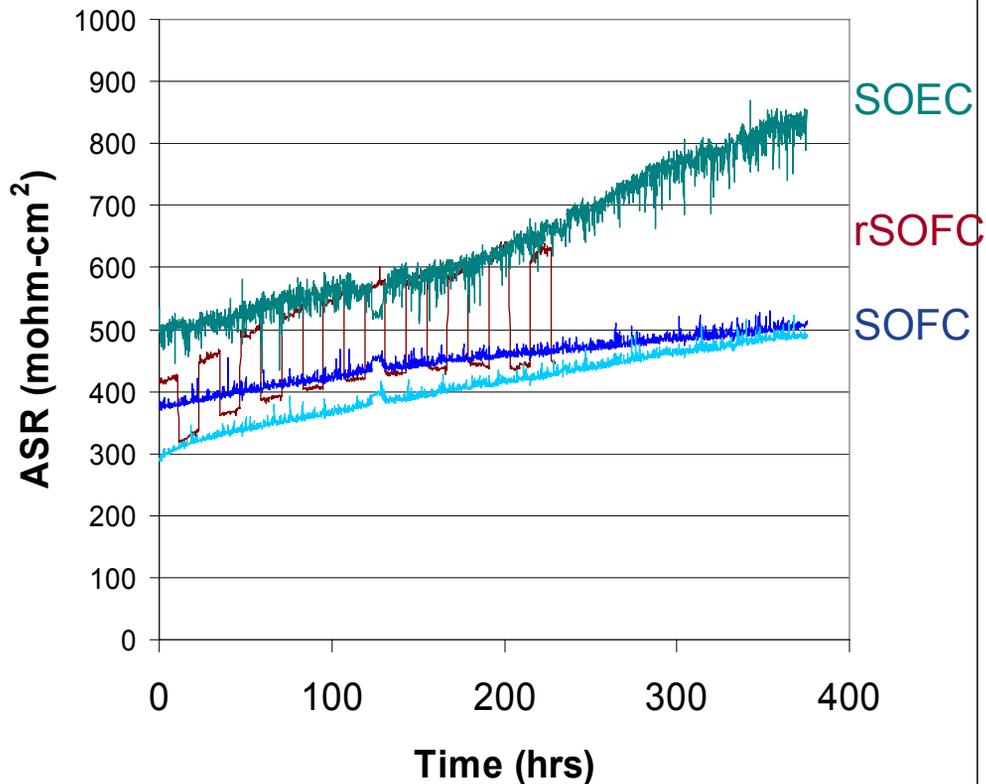


- Vacancy diffusion and activation at the oxygen electrode/electrolyte interface are different for fuel cell mode and electrolysis mode
- Higher current densities can lead to depletion of vacancies at the interface in electrolysis mode
- Experimental data matched well with non-symmetrical vacancy model



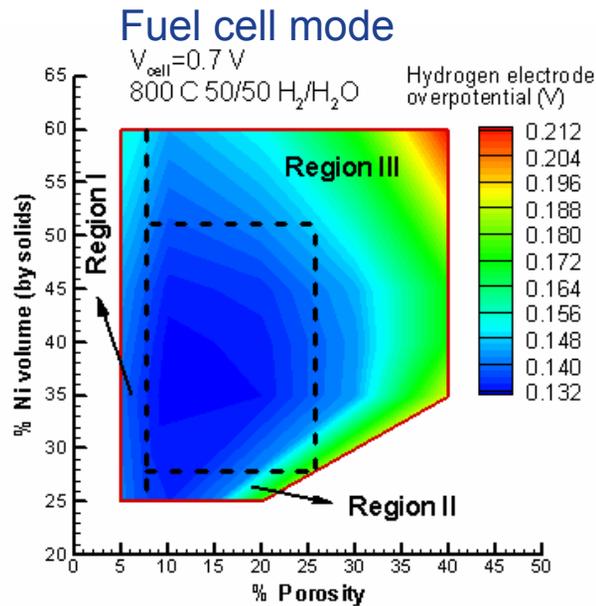
# Operation Mode Cyclic Ability

800 C 50%H<sub>2</sub>O/50%H<sub>2</sub>



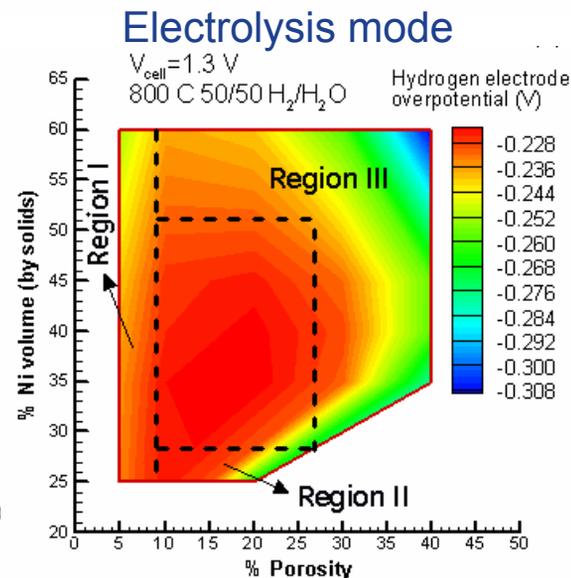
- Evaluated cell performance for fuel cell mode alone, electrolysis mode lone, and fuel cell/electrolysis cyclic mode
- Similar degradation in fuel cell (SOFC), electrolysis (SOEC) and cyclic modes (rSOFC) – perhaps enhanced electrolysis degradation

# Hydrogen Electrode Performance



- Higher polarization losses predicted under electrolysis mode mainly due to difference of diffusion

- Thinner electrode and smaller particles preferred



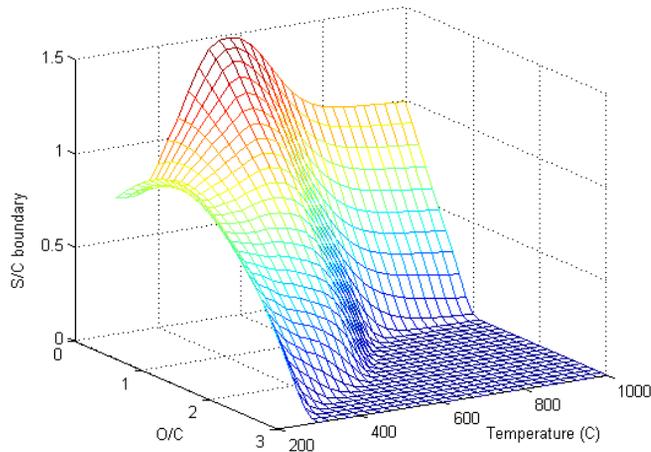
**Conditions:**  
 $T = 800 \text{ C}$   
 Fuel = 50/50  $\text{H}_2/\text{H}_2\text{O}$   
 Active layer thickness =  $16 \mu\text{m}$   
 Active layer particle size =  $0.8 \mu\text{m}$

Region I –  $\text{H}_2/\text{H}_2\text{O}$  diffusion and reaction limited  
 Region II – Reaction limited  
 Region III – Ion conduction and reaction limited

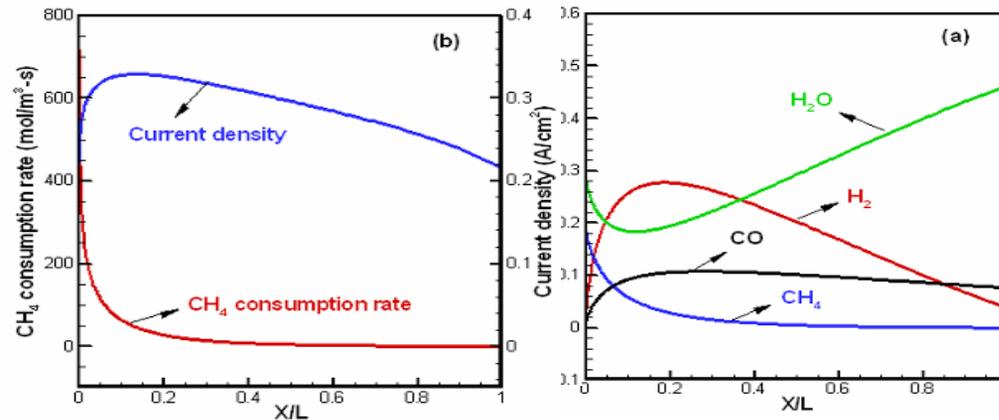


# Hydrogen Electrode Internal Reforming

Thermodynamic Prediction of Carbon Deposition Boundary

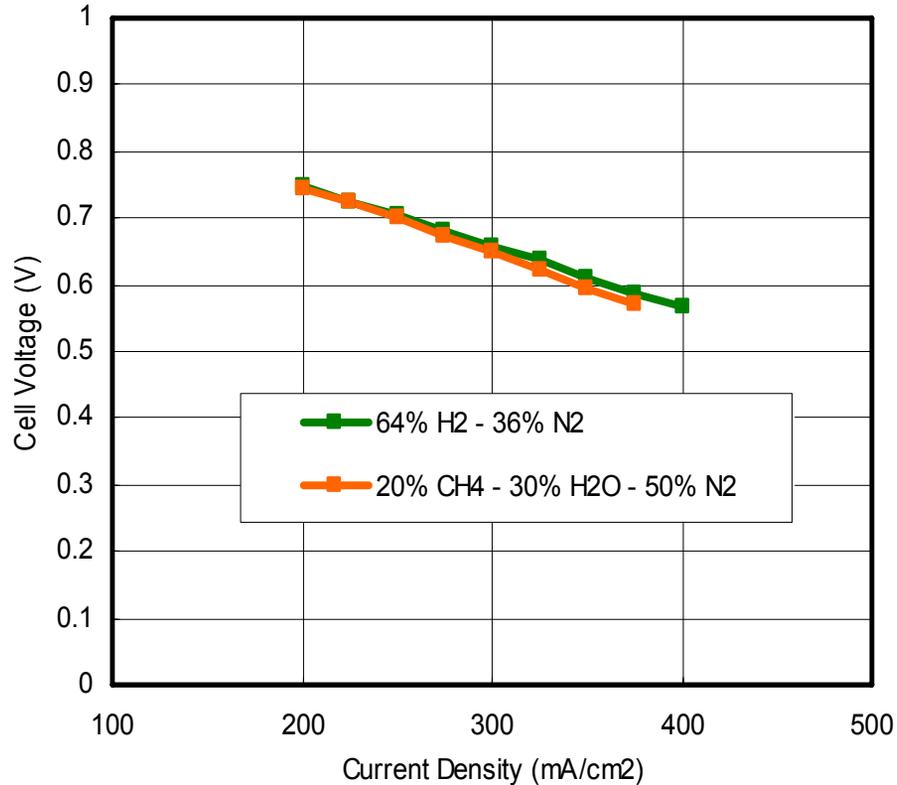


- At 800°C, internal reforming kinetic was fast
- CH<sub>4</sub> conversion measured (gas chromatography) > 98%, agrees well with thermodynamic prediction
- Thermodynamic calculations defined carbon deposition boundary



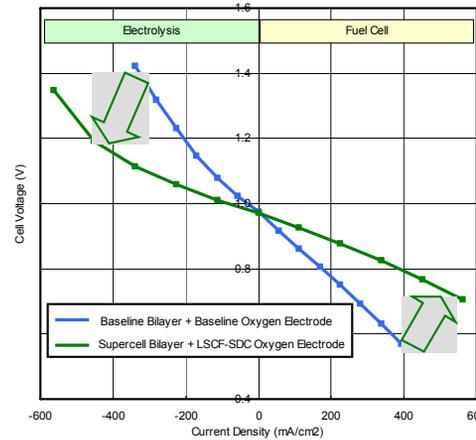
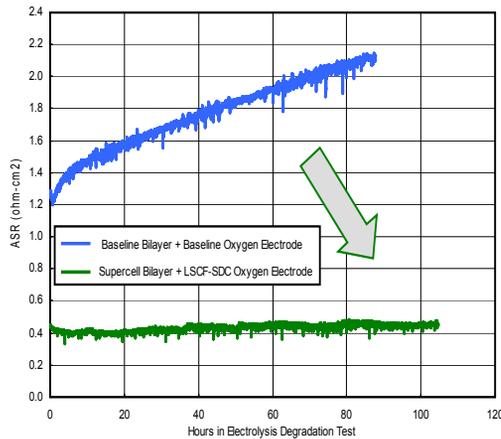
X is the distance from the fuel inlet along the channel and L is the total channel length

# Performance with Internal Reforming

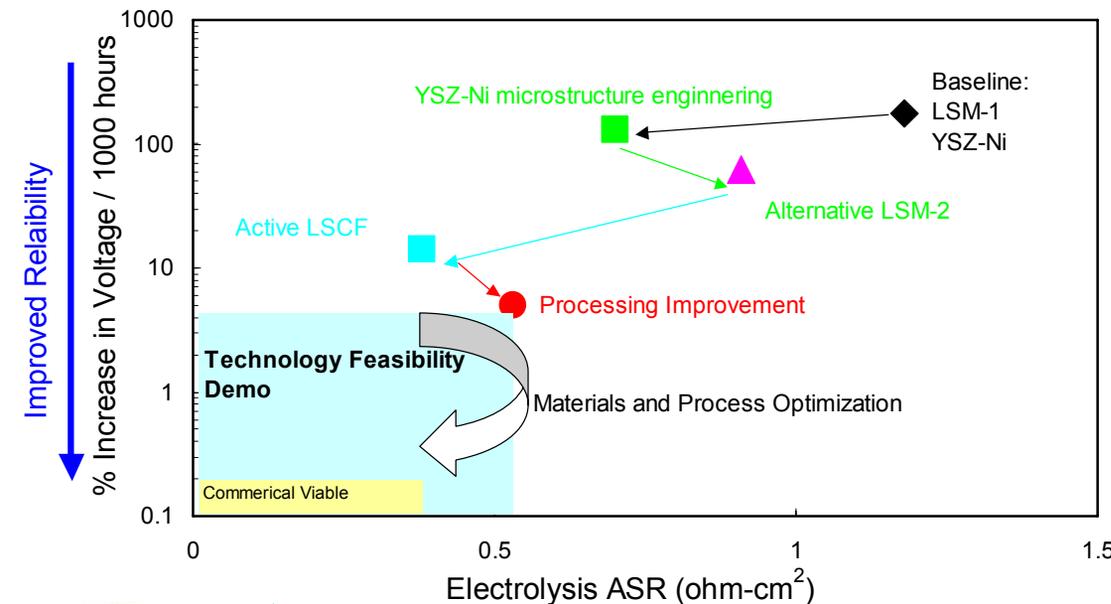


- Performance (I-V curve) with internal reforming similar to that with 64% H<sub>2</sub>/36%N<sub>2</sub> fuel
- Improved cells efficiency and potential system simplification with internal reforming

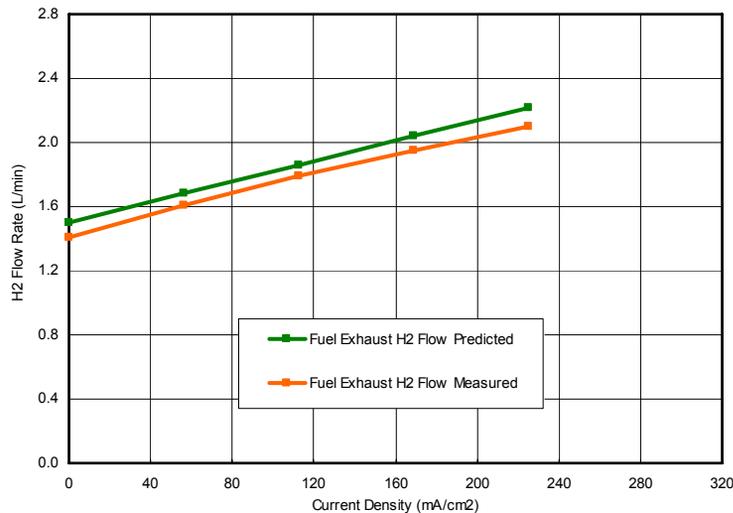
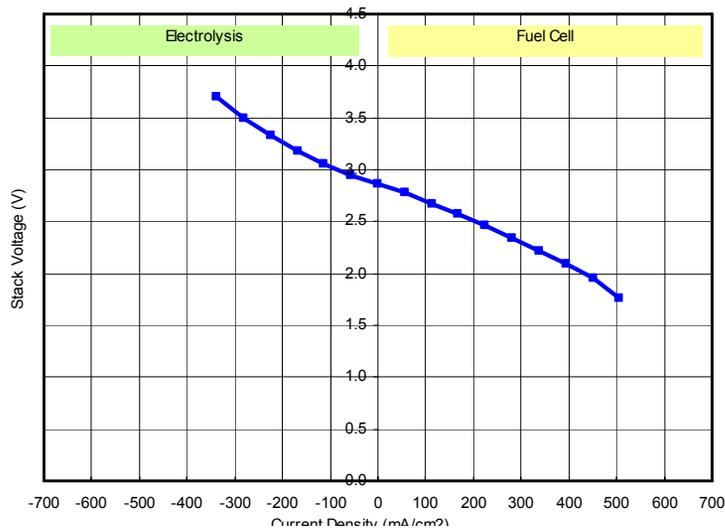
# Module Performance Improvement



- LSCF performed better than LSM/YSZ electrode
- Substantial degradation rate reduction achieved with LSCF oxygen electrode in electrolysis mode
- Improved performance with electrode material selection and process engineering



# Multi-cell Stack Performance



- Built and tested 3-cell stacks under power generation and electrolysis mode for more than 1000 hrs
- Hydrogen production measured and the measured value close to the predicted
- Cell-cell performance variation needs to be addressed

# Future Works

- Demonstrated multi-cell stack operation and assess performance under reversible operating conditions
- Estimate hydrogen production cost (\$/kg H<sub>2</sub>)
- Conduct technology assessment and gap analysis

# Summary

- **Oxygen electrode development**

- Performance: LSCF>LSF>LSM
- “Irreversibility” of oxygen electrode observed, associated with differences in vacancy diffusion and activation at electrode/electrolyte interface

- **Hydrogen electrode development**

- Internal reforming with Ni-YSZ modeled and demonstrated
- Higher polarization loss under electrolysis mode expected, mainly due to difference of H<sub>2</sub> and H<sub>2</sub>O diffusion

- **Module and stack development**

- Module and stack performance improved by electrode engineering
- Initial multi-cell stacks tested and hydrogen generation demonstrated

# Acknowledgement

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