



2013 Hydrogen Program Annual Merit Review Meeting

PEM Electrolyzer Incorporating an Advanced Low Cost Membrane

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Project ID# PD030

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Overview

Timeline

- **Project Start:** May 2008
- **Project End:** April 2013
- **Percent Complete:** 100

Budget

- **Total Project Budget:**
\$2.49MM
 - **DOE Share:**
\$1.99MM
 - **Contractor Share:**
\$0.51MM
- **Funding Received
in FY12:** \$278K

Barriers

Hydrogen Generation by Water Electrolysis

- **G. Capital Cost**
- **H. System Efficiency**

Technical Targets: Distributed Forecourt Water Electrolysis¹

Characteristics		Units	2015	2020	Giner Status (2013)
Hydrogen Levelized Cost ²		\$/kg-H ₂	3.90	<2.30	3.64 ³ (5.11) ⁴
Electrolyzer Cap. Cost		\$/kg-H ₂	0.50	0.50	1.30 (0.74) ⁵
Efficiency	System	%LHV (kWh/kg)	72 (46)	75 (44)	65 (51)
	Stack	%LHV (kWh/kg)	76 (44)	77 (43)	74 (45)

¹ 2012 MYRDD Plan. ² Production Only. ³ Utilizing H2A Ver.2. ⁴ Utilizing H2A Ver.3 (Electric costs increased to \$0.057/kWh from 0.039\$/kWh). ⁵ Stack Only

Partners

- **Parker Hannifin Corporation (Industry)**– System Development
- **Virginia Tech University (Academic)**– Membrane Development

Collaborations

- **3M Fuel Cell Components Program**– NSTF Catalyst & Membrane
- **Entegris** – Carbon Cell-Separators
- **TreadStone Technologies** – Metal Cell-Separators
- **Tokuyama** – Low-Cost Membrane
- **Prof. R. Zalosh (WPI)** – Hydrogen Safety Codes

Relevance: Project Objectives

Overall Project Objectives

- Develop and demonstrate advanced low-cost, moderate-pressure PEM water electrolyzer system to meet DOE targets for distributed electrolysis.
 - Develop high efficiency, low-cost membrane
 - Develop long-life cell-separator
 - Develop lower-cost prototype electrolyzer stack & system

Relevance

- Successful development of a low-cost hydrogen generator will enable
 - Integration of renewable energy sources
 - Early adoption of fuel cell vehicles

FY 2012-13 Objectives

- Deliver/demonstrate prototype electrolyzer system at NREL
- Complete membrane evaluations under aggressive conditions
 - High pressure evaluation
 - High current density evaluation

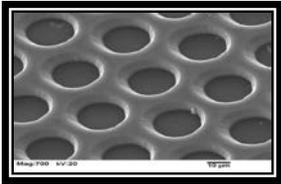


**Low-Cost
PEM
Electrolyzer
Stack**

Approach: Overview

Membrane

- Develop High-Strength, High-efficiency membranes



DSM

- DSM-PFSA ionomer incorporated in an engineering plastic support
- Investigate Alternative Low-Cost Membranes
 - Hydrocarbons ionomers
 - Bi-Phenyl Sulfone (VT)
 - PFSA (850EW) membrane (3M)
- **2012-2013:** Evaluate membrane under aggressive conditions

Cell-Separator

- Develop cell-separators with
 - High electrical conductivity
 - Resistant to hydrogen embrittlement
 - Stable in oxidizing environment
 - Low-Cost



- Evaluate methods of bonding dissimilar metal films
- Evaluate non-metal substrate with conductive coating
- **2012-2013:** Investigate alternative cell-separator materials for future cost reductions

Electrolyzer Stack

- Reduce parts count/cell
- Develop innovative designs to reduce Mat'l costs
- Apply manufacturing methods to reduce costs
- Increase cell active area



- Fabricate 0.5kg-H₂/hr Stack utilizing low-cost components
- **2012-2013:** Broaden product range to include 200 cells/stack

Electrolyzer System

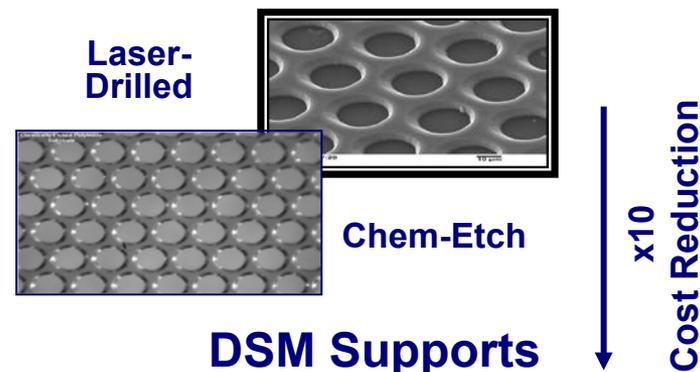
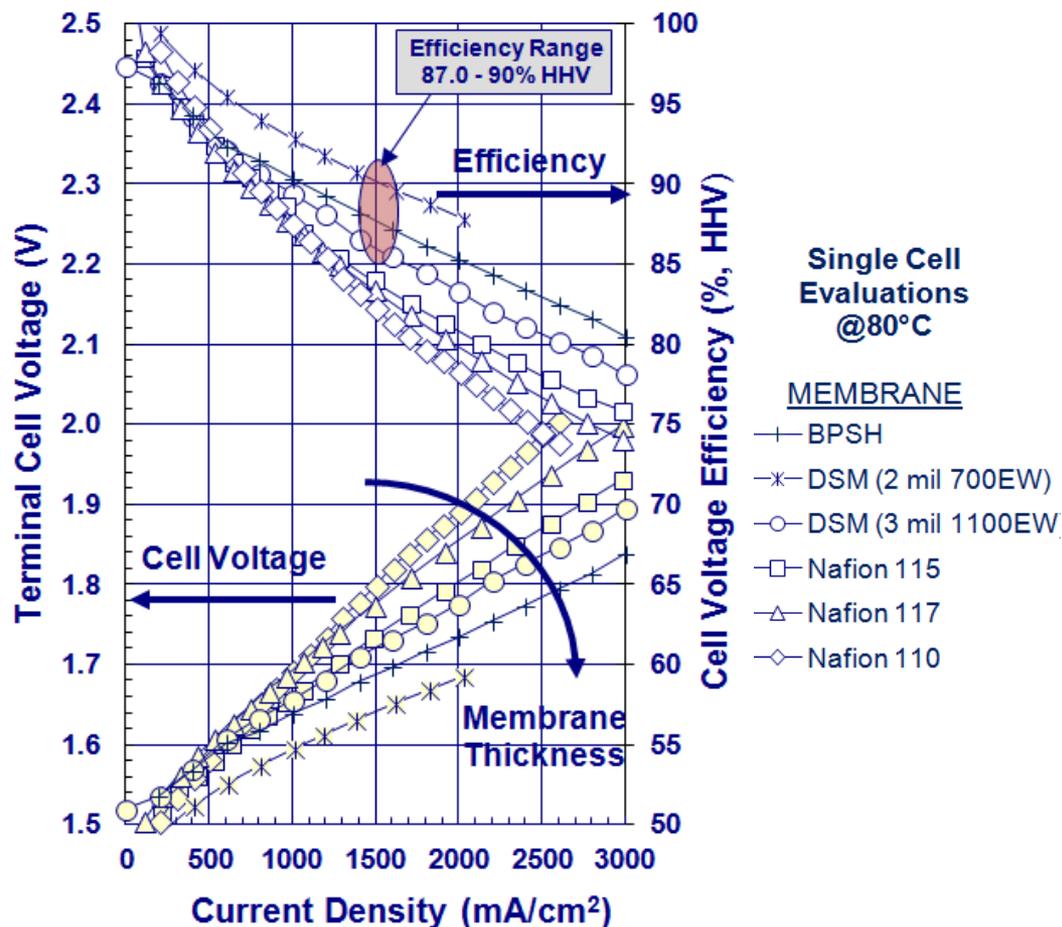
- Reduce BOP capital cost
- Reduce BOP power consumption-through higher efficiency power electronics
- Design high efficiency H₂ dryer
- Improve safety and reliability
- Design for high-volume manufacturing
- Team with large volume commercial manufacturer (**Parker-Hannifin**)



Approach: 2012-13 Milestones

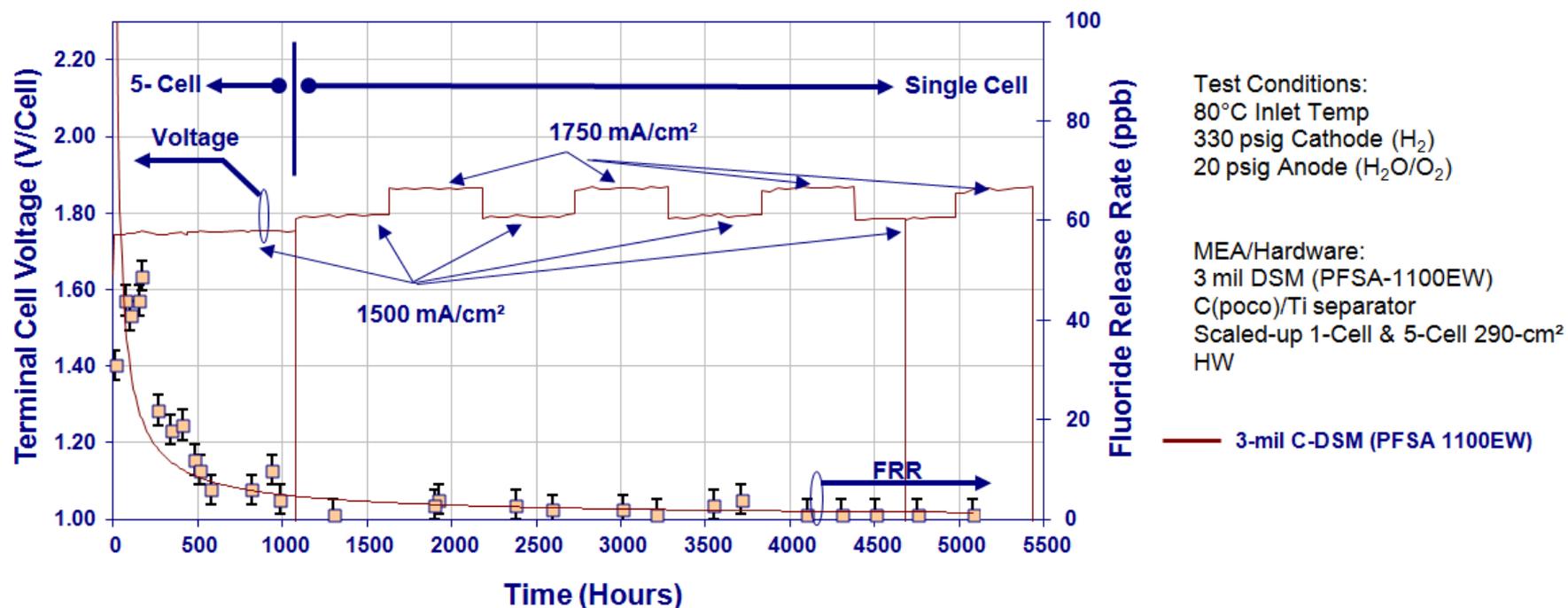
	Go/No Go Decision Points	Progress Notes	%Complete
Membrane	<ul style="list-style-type: none"> Scale-up DSM membrane to 290cm² Evaluated in short stack @ 80°C and 1500-1700 mA/cm² 	<ul style="list-style-type: none"> Operated Scaled-up membrane for 5,000+ hrs Reduced membrane costs via innovative supports Performance DSM > Nafion® 1135 	<p>100% (June 2011)</p>
	<ul style="list-style-type: none"> DSM evaluation at high pressure, high current density 	<ul style="list-style-type: none"> Successfully operated DSM at 5,000 psig Successfully operated DSM at 5,000 mA/cm², 1,000 hrs 	<p>100% (Mar. 2013)</p>
Cell Separator	<ul style="list-style-type: none"> Evaluate cell-separators in short stacks @ 80°C for 5,000 hrs 	<ul style="list-style-type: none"> Completed investigation of new Mat'l for future cost reductions. Includes: nitrided components, low-cost carbon (Entegris), and TreadStone cell-separators Testing Completed – 5,000+ hrs Projected cell-separator lifetime: 60,000+ hrs 	<p>100% (Sep. 2012)</p>
Stack/System Development	<ul style="list-style-type: none"> Completed fabrication of prototype electrolyzer system capable of providing 12 kg-H₂/day at 300-400 psi that has the potential of meeting DOE's cost target for distributed H₂ production 	<ul style="list-style-type: none"> System delivered to NREL and validated Completed DOE's Joule Milestone 	<p>100% (June 2012)</p>

Membrane Progress: Membrane/Catalyst Evaluations



- Developed high efficiency DSM membranes
 - Chem-etched substrates used to lower cost, aid ease of fabrication
- Developed electrode structures with reduced catalyst loadings: 0.7 mg Pt/cm² (Pt/Ir-Anode), 0.4 mg Pt/cm² (Pt/carbon-Cathode)
 - Previously 8 mg Pt/cm²
- Successful testing of 3M NSTF Pt (cathode) and PtIr (anode) catalyst: 3M catalysts are one-order magnitude lower (~0.10 to 0.15 mg Pt/cm² Anode/Cathode)
- Alternative BPSH hydrocarbon membranes exhibited high degradation rates but are effective in reducing cross-over

Membrane Progress: Durability Testing (5,000 hours)



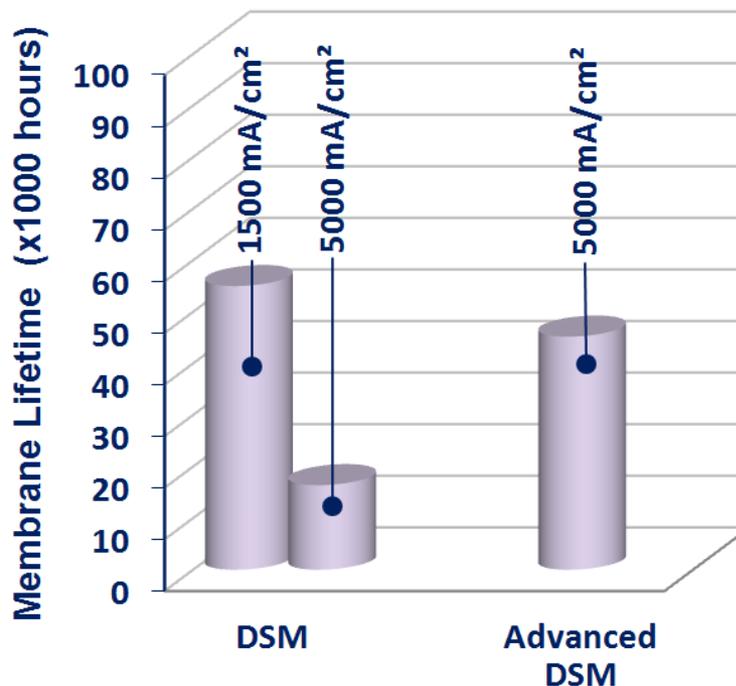
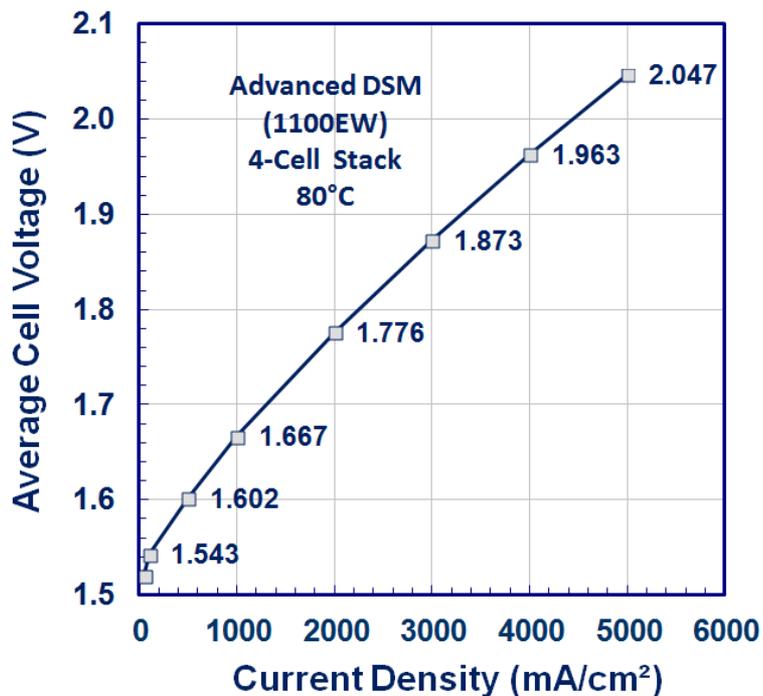
Performance

- Completed 1,000 & 5,000 Hour Life-Test Milestones
 - Scaled-up 5-cell (290-cm²)
 - 1.73-1.75V (~88% HHV)
- DSM MEA from 5-cell short stack re-assembled into a single-cell stack, total operating time = 5430 hours
- Scaled-up cells include low-cost components used in final stack assembly

Membrane Degradation (Estimated Lifetime)

- F ion Release Rate: 3.7 $\mu\text{g/hr}$ (<10 ppb)
- DSM -1100EW (Stabilized Ionomer): ~55,000 hours

Membrane Progress: High Current Density Operation (5,000 mA/cm²)



Performance @ 5,000 mA/cm²

- Advanced DSM: Improved membrane stability at high operating current density
- Operated 4-Cell stack at 5,000 mA/cm² for 1,000 hours
 - Average cell voltage: 2.05V (~74% HHV)
- DSM can endure operation at 10,000 mA/cm² (in continuous 24 hour test)

Membrane Degradation (Estimated Lifetime)

- In PFSA membranes, high current density will reduce lifetime
- Advanced DSM with proprietary additive mitigates degradation and improves life at high operating current densities
 - 200,000 hour lifetime expected at 1,500 mA/cm²

Membrane Progress: Mechanical Stability and High pressure Operation (5,000 psig)



Hydrogen at 5,000 psig
(Ambient O₂)
Generated directly in
PEM Electrolyzer

- DSM Utilized in High-Pressure Operation
 - Stack components developed under current program used in the fabrication of high-pressure stacks
 - Utilizes containment rings, eliminates need for stack enclosure (or external support dome)

DSM Membrane

- High-strength
- No x-y dimensional changes upon wet/dry or freeze-thaw cycling
- Customized MEAs provide more support at edge regions and/or at ports under extreme clamping loads
- Demonstrated significant improvement in membrane creep property and mechanical stability

5,000 psi Giner Electrolyzer
Multi-Cell Stack Design
Work conducted under DOE Program DE-
SC0001486 (see Poster PD065)

Cell-Separator Progress



Carbon/Titanium

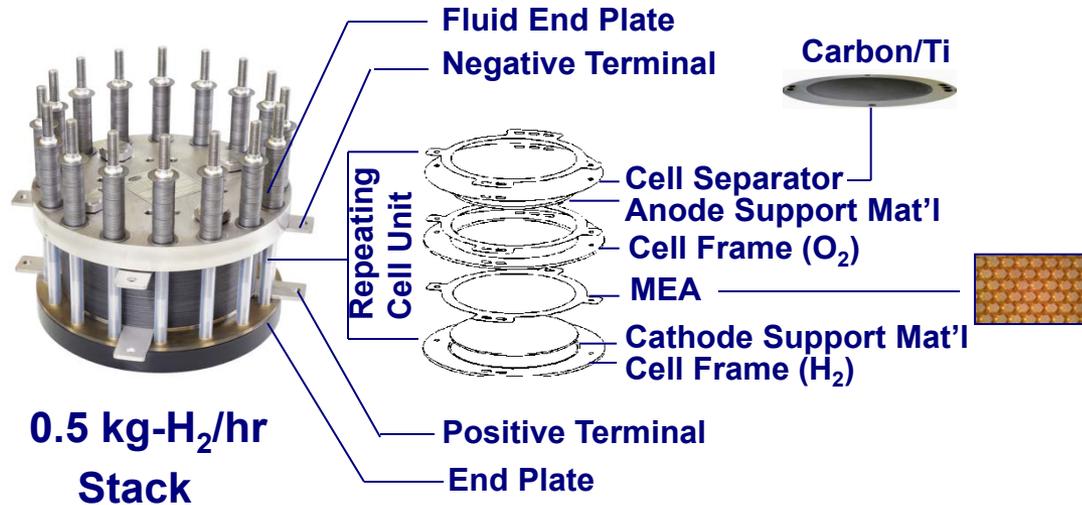
- Carbon/Titanium Cell-Separators Scaled-up to 290-cm²
 - Cell-Separators fabricated with low porosity carbon
 - POCO Pyrolytic Graphite (Surface Sealed)
 - Evaluated in short stack for 5,000+ hours
 - Utilized in final stack build
- Analysis
 - Low hydrogen uptake (low embrittlement)
 - **Lifetime estimate of C/Ti Cell Separators > 60,000 hours**
- Alternative low-cost materials identified
 - Low-Cost Carbon, Nitrided, & TreadStone Cell-Separators
 - Zr/Ti & ZrN/Ti indicate long lifetime, but loss of coating

Cell -Separator	Time (Hours)	H ₂ uptake (ppm)
C/Ti (290-cm²)	5430	104
Zr/Ti(160-cm ²)	500	140
ZrN/Ti (160-cm ²)	500	31
Dual Layer Ti (160-cm²)	500	1105
Ti (baseline)	0	≈ 60
Ti Failure/Embrittlement: ~8000 ppm		

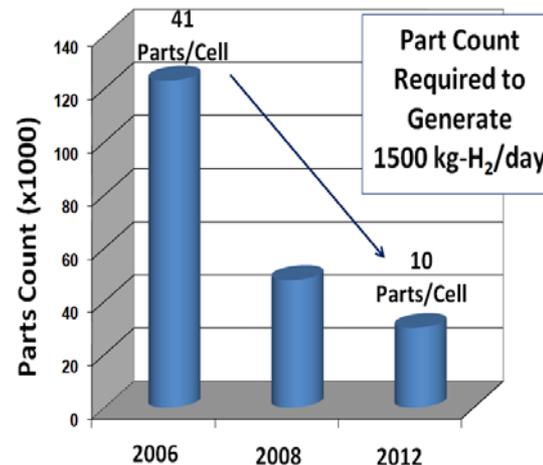
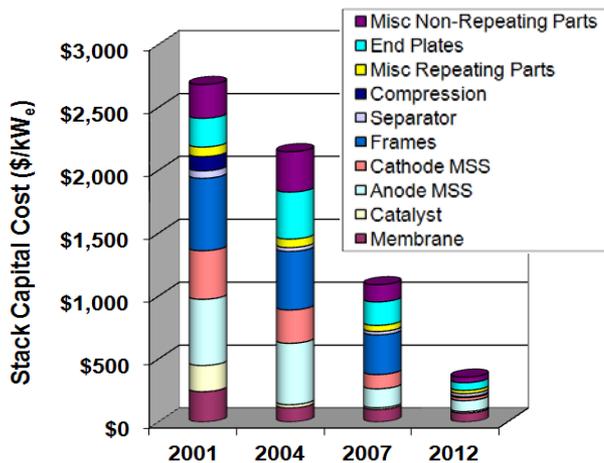
Property	Units	DOE Target FC Bipolar Plates 2017 ¹	GES C/Ti Cell-Separator 2012
Cost	\$/kW	3	> 10
Electrical Conductivity	S/cm	> 100	>300 (680 Poco)
Flexural Strength	MPa	>25	86.1 (Poco)

¹Fuel Cell Technologies Office 2007 MYRD&D

Stack Progress: Advancements & Cost Reductions



The repeating cell unit comprises 90% of electrolyzer stack cost



Stack Improvements

>60% Stack Cost Reduction

- Increased active area (160->>290cm²)
- Reduced catalyst loadings 8->1 mg/cm²
- Reduced Part Count from 41 to 10 Parts/Cell-50% labor reduction
- Pressure Pad: Sub-assembly eliminated
- Molded Thermoplastic Cell Frame
- Cell-Separators: Replaced Nb/Ti with Carbon/Ti
- Frame Thickness reduced (by 30%)
 - Reduces Cathode & Anode Support Mat'l
- DSM MEAs fabricated w/chem-etch supports- 90% cost reduction
- Carbon Steel End Plate (previously S.S.) - 66% material cost reduction
- **Stack commercialized**
- **Broadened product range to include large multi-cell stacks(200+ cells/stack)**
- **CE Compliant**

System Progress



- Assembly: 100% Complete
- Completed series of manuals covering construction, safety and performance
 - System Training Manual
 - System Operation Manual
 - Hydrogen Safety & Response Plan
 - Failure Modes and Effects Analysis (FMEA)
- System delivered to NREL for validation

System Specs:

System oversized to accommodate larger stacks.

Dimensions: 7.2' tall x 6.6' long x 7.8' wide.

3 Compartments (H₂, O₂, and Power Supply/Controls)

Production Rate

0.5 kg H₂/hr (-3.4% dryer)

2.0 kg-H₂/hr (w/ larger Stack & Power Supply)

Operating Pressure

H₂ 390 psig; O₂ atm

Operating Temperature

80°C

Membrane

DSM-PFSA,

Stack Size

Utilized low-cost stack (290 cm²/cell, 27 Cells)

Stack Current Density

1500-1900 mA/cm²

Other

Water Consumption: 5.75 liters/hr

Max. Stack Power Requirement: 24 kW

Heat Rejection: 3.3 kW

Dual-column dryer to reduce maintenance and desiccant replacement

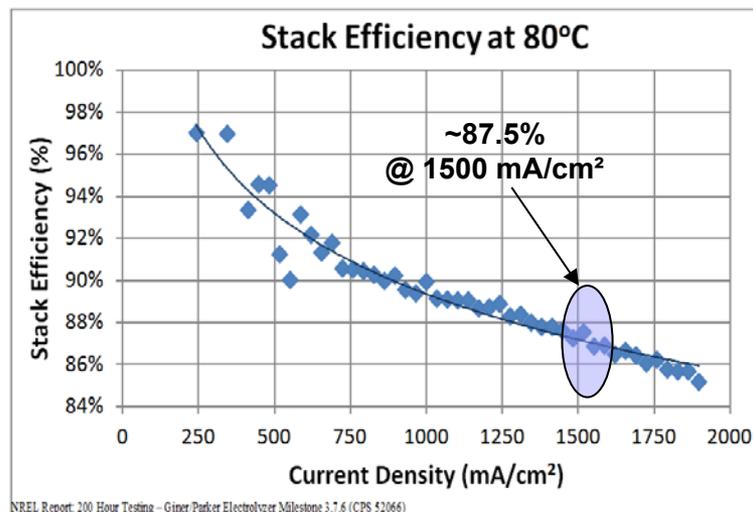
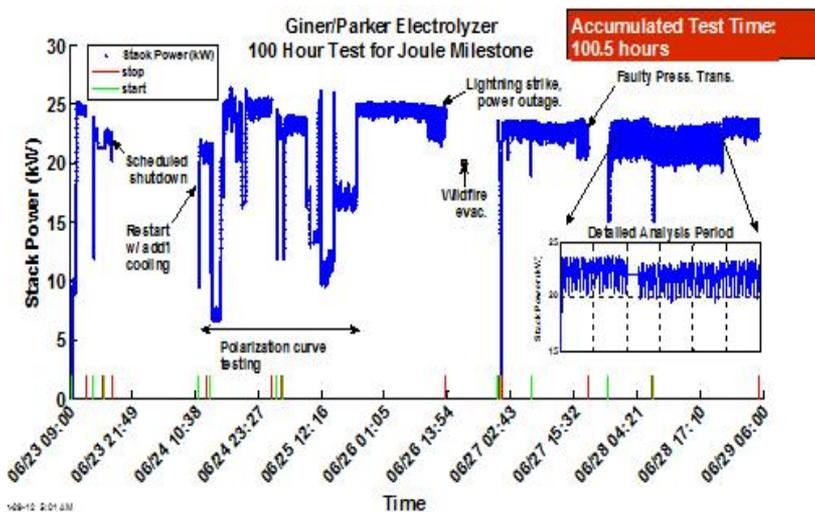
System Enhancements:

- Eliminated stack enclosure (Dome)
- Added ventilation fan to satisfy safety Hydrogen Refueling System Safety Codes
- Electrical lockouts added to stack compartment

System Progress: Validation



- System Validated at NREL in June 2012
 - Nominal operating conditions: 390 psig, 1500 -1900 mA/cm²
 - High stack voltage efficiency: >87% HHV (73.6% LHV) @ 1500 mA/cm²; Energy efficiency; 46.6 kWh_e/kg-H₂
 - Stack Efficiencies in line with DOE 2012 goals
 - Hydrogen drying: 3.4%
 - DOE Joule Milestone Completed !
-
- 3rd party validation of stack by Areva
 - Operating at multiple sites
 - Customer confirms 2,000+ hours at 47 kWh_e/kg @ 1,700 mA/cm²



Giner/Parker Validation

Hydrogen Production & Losses	Units	1500 mA/cm ²	1750 mA/cm ²	1900 mA/cm ²
Stack H ₂ -Production	kg-H ₂ /hr	0.445	0.519	0.563
Membrane permeation losses (-0.6%)		-0.003	-0.003	-0.011
Phase-Separator (-0.14%)		-0.0006	-0.0007	-0.0011
H ₂ -Dryer (3 to 4%)		-0.018	-0.021	-0.022
Total H₂-Production		0.424	0.494	0.529

Power Consumption	Units	1500 mA/cm ²	1750 mA/cm ²	1900 mA/cm ²	
Electrolyzer Stack	kW	20.6	24.2	27.0	
DC power supply & control (assuming 94% eff.)		+1.23	+1.45	+2.3	
PLC Rack		0.05	0.05	0.05	
Electrolyzer Water Pump		0.30	0.30	0.30	
Heat exchanger fans A & B		0.05	0.05	0.05	
H ₂ sensor circuit pump		0.12	0.12	0.12	
Total Energy Consumption (No Dryer)		22.3	26.2	29.82	
H ₂ -Dryer		Chiller (1.4kW Max)	0.46	0.60	0.82
		Heaters A & B	0.07	0.07	0.07
Total Power Consumption (w/Dryer)	22.9	26.8	30.71		

Overall Efficiencies	Units	1500 mA/cm ²	1750 mA/cm ²	1900 mA/cm ²
Electrolyzer Stack (includes permeation)	kWh/kg	46.6	46.9	48.9
System (No Dryer)		50.5	50.8	54.1
System (w/Dryer)		54.0	54.2	58.0

*NREL DATA

1600* mA/cm²
0.468
-0.005
-0.0007
-0.015*
0.43*

1600* mA/cm²
21.9 ± 3.3*
+ 4.2
0.05
0.30
0.05
0.12
26.6 (+0.7)
0.52
0.07
27.9 ± 3.8*

1600* mA/cm²
47.3
57.5 (+1.6)
64.8*

Operating Range:
1300-1800 mA/cm²

H₂-Dryer Losses:
3.4%

Near Theoretical
of 0.44 kg-H₂/hr

Off-the-shelf
Power Supply
Efficiency was Low
(Large Forecourt
Rectifiers >95%)

Includes 0.7kW for
Safety Ventilation
Fans (or +1.6
kWh/kg)

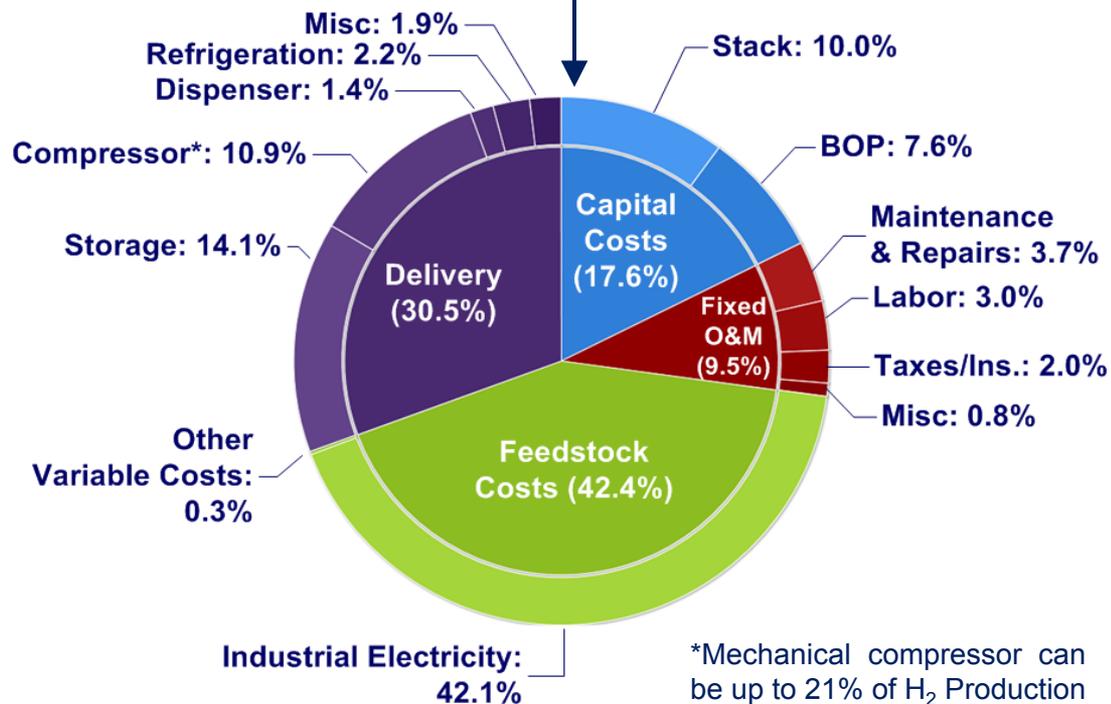
~9 kWh/kg loss
due to power
supply & safety
ventilation

Projected H₂ Cost

H2A Forecourt Model Analysis

H ₂ Production Cost Contribution	H2A Ver. 2.1.1	H2A Ver. 3.0
	(FY 2012)	(FY 2013)
Capital Costs	\$1.06	\$1.30
Fixed O&M	\$0.59	\$0.70
Feedstock Costs @ Efficiency: 50.5 kWh _e /kg -H ₂	\$1.97 (\$0.039/kW)	\$3.09 (\$0.057/kW)
Other Variable Costs (including utilities)	\$0.01	\$0.02
Total Hydrogen Production Cost (\$/kg)	3.64	5.11
Delivery (CSD)	\$1.80 (300 psig output)	\$2.24 (600 psig output)
Total Hydrogen Production Cost (\$/kg)	5.43	7.35

Design Capacity: 1500 kg H₂/day. Assumes large scale production costs for 500th unit

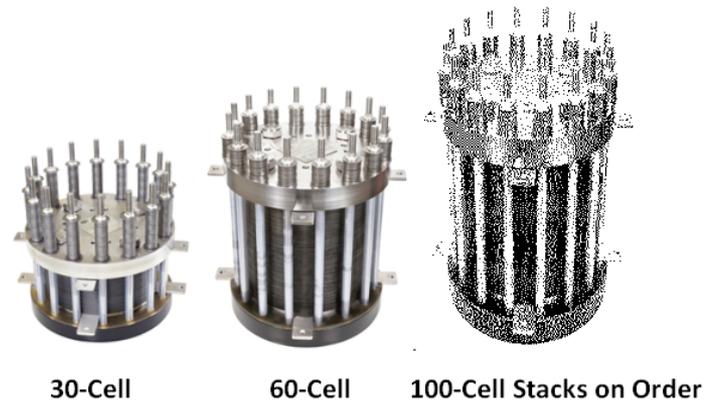


*Mechanical compressor can be up to 21% of H₂ Production Cost (70% of Delivery) depending on maintenance

- Industrial electricity at \$0.039/kWh in H2A Ver. 2, \$0.057/kWh; H2A Ver. 3 (in addition to higher installation & delivery costs)
 - FY2012: Stack output 333 psig H₂ (compressed to 6,250 psig) FY2013: Stack output 600 psig H₂ (compressed to 12,688 psig)
- **Progress inline with achieving new 2015 Target of \$3.90/kg-H₂**

Summary

- **Membrane: Demonstrated Reproducibility, Durability, and Efficiency**
 - Demonstrated high efficiency DSM membranes (single-cell, 5-cell, and 27-cell stacks)
 - Demonstrated 5,000+ hrs lifetime of scaled-up (290 cm²) DSM membrane at 80°C
 - Demonstrated high current density (5,000 mA/cm²) and high pressure (5,000 psig) operation
 - Cell voltage efficiency >87%HHV, 46.6 kWh_e/kg-H₂ @ 1500 mA/cm² meeting 2012 DOE targets
- **Cell Separator & Component Development:**
 - Demonstrated 5,000+ hrs lifetime of scaled-up cell-separators
 - Demonstrated significantly reduced hydrogen embrittlement with carbon/Ti and TreadStone cell-separators
 - Expected cell-separator lifetime range: 60,000+ Hours
- **Scaled-Up Stack:**
 - Significant progress made in stack cost-reduction (cell-components, membrane, & catalyst)
 - 60% reduction in stack cost
 - Stack Commercialized & In production :30, 60, and 100-cell configurations
- **System Development:**
 - Prototype system delivered to NREL
 - DOE Joule Milestone completed
 - Negotiating with multiple OEMs, “Giner-Inside” branded systems



Future Plans & Challenges: FY2013 and Forward

2020 cost targets require further cost reductions and improvements in efficiency

Membrane

- Improve membrane performance
 - Higher operating temperatures, pressures, and current densities required to meet new targets
 - Lower EW ionomers
 - Reduce membrane resistance
 - Improve chemical stability

Stack

- Reduce Stack Costs
 - Labor is 33-50% cost of stack
 - Reduce labor cost through new manufacturing techniques
 - New low-cost materials
 - Reduce part count in cells
 - Unitize cell components (to further reduce parts/cell)
 - Increase stack active-area to 1ft² (or larger) for large energy storage applications

Distributed Forecourt Water Electrolysis ¹	
H ₂ Production Cost Contribution	New DOE Target (2020)
Capital Costs	\$0.50
Fixed O&M	\$0.20
Feedstock Costs @ Efficiency: 50.5 kWh _e /kg -H ₂	\$1.60 (46.9kWh/kg) (\$0.037/kW)
Other Variable Costs (including utilities)	<\$0.10
Total Hydrogen Production Cost (\$/kg)	2.30
Delivery (CSD)	\$1.70
Total Hydrogen Production Cost (\$/kg)	<4.00

¹2012 MYRDD Plan

Future Plans & Challenges...

Stack (Cont...)

- Design for automated assembly
- Improve pressure capabilities of stacks (6,250 and 12,688 psi)
- Demonstrate stacks under aggressive conditions (wind to hydrogen applications)
 - High current density operation

System

- Simplify electrolyzer systems to reduce cost
- Unify BOP components
 - In Regenerative Fuel Cell Systems, combine subsystems

Validation

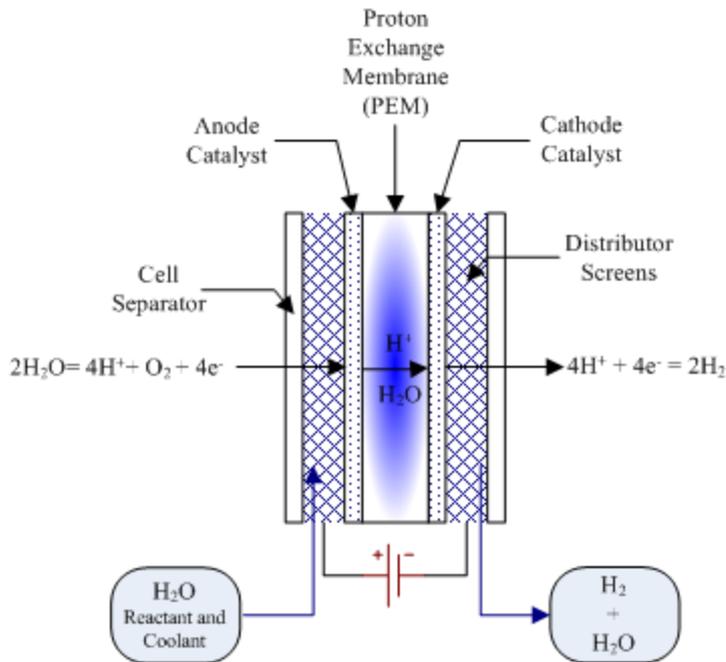
- Industrial collaborations needed to promote technology
- Testing facilities for validation of large MW scale electrolyzers are needed

AMR

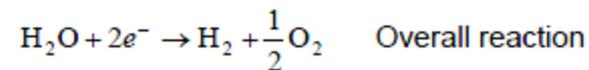
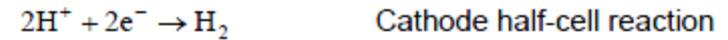
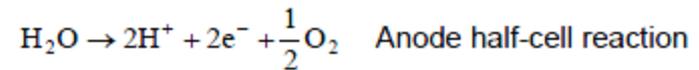
Technical Slides

Technical Slide 1-

Technical principle of the PEM-based water electrolysis



PEM Cell Reactions



Water permeation through PEM

$$\sim 3\text{H}_2\text{O}/\text{H}^+$$